Folate Antagonists. 6. Synthesis and Antimalarial Effects of Fused 2,4-Diaminothieno [2,3-d] pyrimidines (1-3)

Edward F. Elslager, * Patricia Jacob and Leslie M. Werbel

Chemistry Department, Research and Development Division, Parke, Davis and Company, Ann Arbor, Michigan 48106

Received April 10, 1972

2,4-Diamino-5,7-dihydro-6*H*-thiopyrano[4',3':4,5]thieno[2,3-*d*]pyrimidine, 2,4-diamino-9*H*-indeno[1',2':4,5]thieno[2,3-*d*]pyrimidine, 2,4-diamino-5*H*-indeno[2',1':4,5]thieno[2,3-*d*]pyrimidine, 9,11-diamino-5,6-dihydronaphtho[1',2':4,5]thieno[2,3-*d*]pyrimidine, 7,9-diamino-5,6-dihydronaphtho[2',1':4,5]thieno[2,3-*d*]pyrimidine, 2,4-diamino-7-benzyl-5,6,7,8-tetrahydropyrido[4',3':4,5]thieno[2,3-*d*]pyrimidine, and various 2,4-diamino-5,6,7,8-tetrahydro-[1]benzothieno[2,3-*d*]pyrimidines were synthesized by cyclization of the requisite fused 2-aminothiophenene-3-carbonitriles utilizing chloroformamidine hydrochloride in diglyme. Several compounds exhibited strong inhibitory effects against *Streptococcus faecalis* (MGH-2), *Staphylococcus aureus* (UC-76), *Streptococcus faecium* (ATCC 8043), *Lactobacillus casei* (ATCC 7469), and *Pediococcus cerevisiae* (ATCC 8081) *in vitro*, and three compounds displayed antimalarial activity against *Plasmodium berghei* in mice and *P. falciparum* (Uganda I) *in vitro*.

Contemporary studies on folate antagonists in these laboratories have demonstrated that various 2,4-diamino-6-benzyl-5*H*-pyrrolo[3,4-*d*]pyrimidines (4) and 2,4-diamino-6-benzyl-5,6,7,8-tetrahydropyrido[4,3-*d*]pyrimidines (5) exhibit significant antimalarial, antibacterial, and antimetabolite effects. When administered orally to mice infected with *Plasmodium berghei*, 2,4-diamino-6-(3,4-dichlorobenzyl)-7-methyl-5*H*-pyrrolo[3,4-*d*]pyrimidine (I) was equiactive with quinine hydrochloride (4),

while 2,4-diamino-6-(3,4-dichlorobenzyl)-5,6,7,8-tetrahydropyrido [4,3-d] pyrimidine dihydrochloride (II) was approximately twenty times as potent as this reference drug (5). The latter compound (II) was more active than the corresponding quinazoline antifolate 2,4-diamino-6-[(3,4-dichlorobenzyl)amino]quinazoline IIIa ((2,3,4)-dichlorobenzyl)amino]-(3,4-dichlorobenzyl)amino]-

pyrido[2,3-d]pyrimidines IVa (Q = < 0.4) and IVb (Q = 6.2) (8), but was much less potent than other folate antagonists such as 2,4-diamino-6-[(3,4-dichlorobenzyl)nitrosamino]quinazoline (IIIb) (Q = 270) (9,10), 2,4-diamino-6-[(p-chlorobenzyl)isopropylamino]quinazoline (V) (Q = 1160) (11), and 2,4-diamino-6-(2-phenyl-1-pyrrolidinyl)quinazoline (VI) (Q = 210) (12,13).

Certain 2,4-diaminothieno [2,3-d] pyrimidines can be viewed as bioisosteres of the antimalarial 2,4-diamino-5*H*-pyrrolo [3,4-d] pyrimidines (I), 2,4-diamino-5,6,7,8-tetra-hydropyrido [4,3-d] pyrimidines (II), 2,4-diaminopyrido [2,3-d] pyrimidines (IV), and 2,4-diaminoquinazolines (III, V, VI) (1,4-13). Moreover, Roth (14) recently reported that 2,4-diamino-6-benzyl-5-methylthieno [2,3-d] pyrimidine (VII), a thienopyrimidine isostere of the potent

TABLE I

Fused 2-Aminothiophene-3-carbonitriles

Compound Structure	ā	Yield purified, %	Purificn solvent	Procedure	Formula	Carb	Carbon, %	Ana Hydro Calcd.	Analyses Hydrogen, % Calcd. Found	Nitrog Calcd.	Nitrogen, % Calcd. Found
NATURAL STATE OF THE STATE OF T	200.5-209.5	10	MeCIN	-	C8H8N2S2	48.95	48.84	4.11	4.14	14.28	14.20
NO S	142.5.145.5 (a)	61	i-PrOH	- :	$C_9H_{10}N_2S$	60.64	60.64	5.66	5.79	15.72	15.90
	CN 101.0-103		E40H	= -	C12H8N23	68.79	96.70	3.80	4.11	13.20	13.28
\$ \$ \rangle \text{N} \r	cn 165-168	48	C ₆ H ₆	. =	C ₁₃ H ₁₀ N ₂ S		69.00	4.45	4.61	12.38	12.33
NO SHAN S	194-196.5 (b)	40	EtOH	-	C13H10N2S	00.69	69.16	4.45	4.48	12.38	12.49
\$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$ \$, NH2 177.5-180.5	63	EtOH	-	$C_{15}H_{14}N_{2}S$	70.83	70.87	5.55	5.72	11.02	11.20

. .

11.12

05

15.77

9

9 11.0	5.67 15.6
5.7	5.6
5.55	5.61
71.13	92.99
70.83	98.99
30 C_6H_6 II $C_{15}H_{14}N_2S$ 70.83 71.13 5.55 5.76 11.0	C ₁₅ H ₁₅ N ₃ S 66.88 66.76
=	-
C_6H_6	Еюн
30	55
180-183	155-157.5 (c)
$\langle \bigcirc \rangle$	CHP2 S WHP2 CN
œ	တ

(a) Lit. (35) reports m.p. 147-148°. (b) Lit. (30) reports m.p. 195-197°. (c) Lit. (36) reports m.p. 149-152°

antibacterial agent 2,4-diamino-6-benzyl-5-methylpyrido-[2,3-d]pyrimidine (VIII) (15), shows strong inhibitory effects against Lactobacillus casei and, like VIII, displays a favorable inhibition ratio against isolated dihydrofolate reductases from bacterial and mammalian sources (14). More distant relatives, such as the dihydrochloride salts of cis and trans-4-[2-(dimethylamino)ethyl]amino]-6,8-bis(p-chlorophenyl)-5,6-dihydro-8H-thiopyrano[4',3':4,5]-thieno[2,3-d]pyrimidine (IXa) (16), the cis and trans diethyl analogs (IXb) (16), and cis-4-[2-(dimethylamino)ethyl]amino]-6,8-bis(p-chlorophenyl)-7-methyl-5,6,7,8-tetrahydropyrido[4',3':4,5]thieno[2,3-d]pyrimidine (X) (17), are reported to be active against drug-sensitive strains of P. berghei in mice when administered orally in daily

doses of 100 mg./kg. for 4 days. Moreover, compounds of type IXa and b are also active against primaquine-resistant strains of *P. berghei* (16), while X is effective against chloroquine-resistant lines (17). Since these compounds lack a six-membered ring incorporating the sequence -N=C(NH₂)-N=C(NH₂)-, which plays a key role in conferring optimal antiplasmodial effects among folate antagonists (6,18,19), it seems likely that they act by a different mechanism.

We have therefore synthesized several representative fused 2,4-diaminothieno[2,3-d]pyrimidine prototypes for antimalarial evaluation under the auspices of the Walter Reed Army Institute of Research (3). Rosowsky, Modest, and co-workers independently prepared a variety of 2,4-diaminothieno[2,3-d]pyrimidine derivatives, including several of the compounds (11, 15, 16, 19) we had prepared earlier. Preliminary accounts of their studies have already appeared (20,21), and details of their work are being presented in separate communications (22-24).

Chemistry.

Interest in the chemistry of thieno [2,3-d] pyrimidines and fused thieno [2,3-d] pyrimidines has escalated since the first synthesis of a thieno [2,3-d] pyrimidine derivative was reported by Baker and co-workers in 1953 (25). A plethora of thieno [2,3-d] pyrimidines (14,25-30), [1] benzothieno [2,3-d] pyrimidines (27-33), naphtho [2',1':4,5]-thieno [2,3-d] pyrimidines (16), and pyrido [4',3':4,5]-thieno [2,3-d] pyrimidines (17) has been prepared starting

							·		
en, % Found	23.40	25.38	19.24	21.34	20.81	20.32	18.97	18.92	22.12
Nitrog Calcd.	23.51	25.43	19.26	21.92	20.88	20.68	18.91	18.91	22.49
lyses (d) gen, % Found	4.40	5.21	4.86	4.22	4.67	4.70	5.54	5.53	5.61
Anal Hydro, Calcd.	4.23	5.42	5.06	3.99	4.51	4.57	5.42	5.42	5.50
on, % Found	45.29	54.35	58.70	61.48	62.91	62.37	65.04	64.45	61.14
Carbo Calcd.	45.36	54.51	58.61	61.09	62.66	62.04	64.84	64.84	61.71
Formula	C ₉ H ₁₀ N ₄ S ₂	C ₁₀ H ₁₂ N ₄ S	C ₁₃ H ₁₀ N ₄ S·0.6 C ₂ H ₆ O·0.5 H ₂ O (a)	C ₁₃ H ₁₀ N ₄ S-0.07 H ₂ O (b)	C14H12N4S	C ₁₄ H ₁₂ N ₄ S-0.15 H ₂ O (c)	C16H16N4S	C16H16N4S	C1eH17NsS
rocedure	Ξ	≅		E	ž	E	≣	E	H
Purificn solvent F	АсОН	Етон	Егон	DMF-H ₂ O, C ₆ H ₅ CH ₃	ЕғОН	EtO Ac- isooctane	DMF. di NaOH	DMF. dii NH40H	Et0 Ac
Yield purified, %	Ξ	90	10	æ	20	16	29	62	6
	278-282 dec.	241-242.5	> 300	310.312	306-308	257-260	267-269	274-277	178-180.5
Structure	S N N N N N N N N N N N N N N N N N N N	2-HZ 2-HZ 2-HZ 2-HZ 2-HZ 2-HZ 2-HZ 2-HZ	NH2 NH2	z - ż	- Ž	N N N N N N N N N N N N N N N N N N N	S NH2		OF K2 NT2 NT2 NT2
Compound No.	01	E	2	13	15	91	11	85	61
	Yield Analyses (d) purified, Purifien Carbon, % Hydrogen, % Nitros Structure M.p., °C % solvent Procedure Formula Caled. Found Caled.	Yield Analyses (d) purified, purified, purified purified, Purifien Purified purified, Purifien Purified purified, Procedure Formula Calcd. Found	Yield purified, Purified purified, Purified purified, Purified purified, Purified Structure Purified purified, Purified Structure Purified Purified Structure Formula (Caled Found Caled Found Cal	Structure Structure M.p., °C % solvent Procedure Formula Calcd. Found Calcd. F	Structure M.p., °C	Structure M.p., °C % solvent Procedure Formula (Galci. Found Galci. Fo	Structure N. 19, °C Structure Structure N. 19, °C Structure Structure N. 19, °C Structure Structure Structure Structure N. 19, °C Structure Structure	Structure Structure Np. °C 'yield 'yield	Surrature Surrature Surra

• (a) H₂O: calcd., 3.10; found, 3.02. (b) H₂O: calcd., 0.49; found, 0.21. (c) H₂O: calcd., 0.99; found, 0.95. (d) Water analyses by the Karl Fischer Method.

from either thiophene or pyrimidine fragments. However, only one fused 2,4-diaminothieno [2,3-d] pyrimidine has been reported, namely 2,4-diamino-5,6,7,8-tetrahydro-[1] benzothieno [2,3-d] pyrimidine (11) (27). This compound was obtained by the condensation of 2-amino-4,5,6,7-tetrahydrobenzo [b] thiophene-3-carbonitrile with cyanamide and pyridine hydrochloride.

In the present work, a majority of the intermediate fused 2-aminothiophene-3-carbonitriles (XI) (1, 2, 4, 6, 7, 9, Table I) were prepared directly from the corresponding ketones by treatment with sulfur and malononitrile in the presence of morpholine (40-63% yield, procedure I), a modification of the Willgerodt-Kindler reaction of ketones with sulfur and ammonia (34,35). In those instances where the single step procedure was unsatisfactory (3, 5, 8, Table I), a stepwise process was utilized wherein the Knoevenagel-Cope intermediate formed from the ketone and malononitrile was first isolated, and was then treated with sulfur and morpholine or triethylamine (25-48% yield, procedure II). 2-Amino-6-benzyl-4,5,6,7tetrahydrothieno [2,3-c] pyridine-3-carbonitrile (9) and related compounds have recently been synthesized by this overall procedure, and are reported to have analgetic and antiinflammatory activity (36,37).

Cyclization of the intermediate 2-aminothiophene-3-carbonitriles (XI) to the desired fused 2,4-diaminothieno-[2,3-d]pyrimidines (Chart I, Table II) was accomplished utilizing chloroformamidine hydrochloride (38) in diglyme (procedure III), a method developed earlier for the conversion of 2-aminobenzonitriles to 2,4-diamino-6-(hetero-

cyclic)quinazolines (12,13) and 2,4-diamino-6-(aryloxy From compounds 1-4 and aralkoxy)quinazolines (1). and 6-9 (Table I) were thus obtained 2,4-diamino-5,7dihydro-6H-thiopyrano 4',3':4,5 thieno 2,3-d pyrimidine (10) (11%), 2,4-diamino-5,6,7,8-tetrahydro-[1]benzothieno[2,3-d] pyrimidine (11) (30%), 2,4-diamino-9H-indeno-[1',2':4,5] thieno [2,3-d] pyrimidine (12) (10%), 2,4-diamino-5H-indeno[2',1':4,5]thieno[2,3-d]pyrimidine (13) (8%), 7,9-diamino-5,6-dihydronaphtho [2',1':4,5] thieno-[2,3-d] pyrimidine (16) (16%), 2,4-diamino-5,6,7,8-tetrahydro-7-phenyl-[1]benzothieno[2,3-d]pyrimidine (17) (67%), 2,4-diamino-5,6,7,8-tetrahydro-5-phenyl-[1]benzothieno[2,3-d] pyrimidine (18) (62%), and 2,4-diamino-7benzyl-5,6,7,8-tetrahydropyrido[4',3':4,5]thieno[2,3-d]pyrimidine (19) (9%). In the case of 2-amino-4,5-dihydronaphtho[2,1-b]thiophene-1-carbonitrile (5), this condensation afforded the uncyclized intermediate (1-cyano-4,5dihydronaphtho [2,1-b] thien-2-yl) guanidine (14) (procedure IV). Ring-closure to 15 was readily accomplished by heating 14 in dimethylformamide (86%, 50% overall).

The naphtho[1',2':4,5]thieno[2,3-d]pyrimidine (15), indeno[1',2':4,5]thieno[2,3-d]pyrimidine (12), and indeno-[2',1':4,5]thieno[2,3-d]pyrimidine (13) ring systems are unknown based on a search of *Chemical Abstracts* and "The Ring Index" (39), and thus appear to represent novel heterocyclic types.

Biological Results.

Antimalarial Effects.

The fused 2,4-diaminothieno [2,3-d] pyrimidines described in the present communication, together with the condensed 2-aminothiophene-3-carbonitriles employed as synthetic intermediates, were evaluated for antimalarial activity utilizing P. berghei in mice, P. gallinaceum in chicks, and P. falciparum in vitro. The thieno [2,3-d]pyrimidines (10-13, 15-19, Table II) and the 2-aminothiophene-3-carbonitriles [1-5, 7-9 (Table I), and 14] were administered subcutaneously to mice infected with a normal drug-sensitive strain of P. berghei (40,41) in single doses ranging from 20 to 640 mg./kg. 2,4-Diamino-5,7-dihydro-6H-thiopyrano[4',3':4,5]thieno[2,3-d]pyrimidine (10) and 2,4-diamino-5,6,7,8-tetrahydro [1] benzo thieno [2,3-d] pyrimidine (11) each increased the survival time of mice 2.9 days at a dose of 640 mg./kg. This suggests that they may have slight antimalarial activity, although the significance of these data is borderline. Compounds 12, 13, 15, 17, and 18 were tolerated well at the highest levels tested (160-640 mg./kg.), but were ineffective against P. berghei. The thienopyrimidines 16 and 19 were toxic for mice at doses of 160 mg./kg. and 80 mg./kg., respectively, and likewise showed no activity against the plasmodia. Although the intermediate 2aminothiophene-3-carbonitriles (1-5, 7-9, Table I) were tolerated well by mice at 640 mg./kg., no antimalarial activity was observed. The guanidine derivative (14) was toxic at 320 mg./kg. and lacked antimalarial effects.

Three of the carbocyclic 2,4-diaminothieno [2,3-d] pyrimidines (10,13,16) and the intermediate 2-aminothiophene-3-carbonitriles (3,5) were also tested against *P. gallinaceum* infections in white Leghorn cockerels (1,40,41). None exhibited significant anitmalarial effects when given in single intravenous or subcutaneous doses ranging from 10-320 mg./kg., but each was tolerated well.

In contradistinction, 2,4-diamino-7-benzyl-5,6,7,8-tetrahydropyrido[4',3':4,5]thieno[2,3-d]pyrimidine (19) exhibited high activity against the drug-sensitive Uganda I strain of *P. falciparum in vitro* (42-44). When the parasites were incubated with 19 at drug concentrations ranging from 25 to 5,000 μ g./ml., > 90% of the parasites, relative to the control sample, were unable to mature to schizonts. Against the drug-resistant Vietnam Marks strain, however, compound 19 was virtually ineffective over the same concentration range.

Antibacterial Activity.

Three of the condensed 2,4-diaminothieno[2,3-d]pyrimidines (11, 16, 17) were tested in vitro against a spectrum of pathogenic bacteria including Streptococcus faecalis (MGH-2), sensitive (UC-76) and drug-resistant (S18713) Staphylococcus aureus, Pseudomonas aeruginosa (28), Escherichia coli (Vogel), and Shigella sonnei (C-10). A modification of the gradient plate procedure of Szybalski (45) and Webb and Washington (46) was employed throughout (13). 7,9-Diamino-5,6-dihydronaphtho[2',1':4,5]thieno[2,3-d]pyrimidine (16) produced 50% inhibition of S. faecalis (MGH-2) at 1.0 μg./ml. and 2,4-diamino-5,6,7,8-tetrahydro-7-phenyl-[1]benzothieno-[2,3-d] pyrimidine (17) inhibited S. aureus (UC-76) at 25 μ g./ml., but both were inactive against the other bacteria at 25 µg./ml. Compound 11 was inactive against all test organisms at $25 \mu g./ml$.

Antimetabolite Effects.

Three of the thieno[2,3-d]pyrimidines (11, 12, 18) were evaluated for their inhibitory effects on Streptococcus faecium (S. faecalis ATCC 8043), Lactobacillus casei ATCC 7469, and Pediococcus cerevisiae ATCC 8081 (47). 2,4-Diamino-5,6,7,8-tetrahydro[1]benzothieno[2,3-d]pyrimidine (11) and 2,4-diamino-9H-indeno[1',2':4,5]thieno[2,3-d]pyrimidine (12) produced 50% inhibition of S. faecium at 30 and 37 ng./ml., L. casei at 290 and 11,800 ng./ml., and P. cerevisiae at 300 and 185 ng./ml., respectively. Compound 18 was inactive against all three organisms at a concentration of 100,000 ng./ml. Obviously the relative inhibitory activity of the thieno[2,3-d]-pyrimidines against these organisms does not afford a reliable basis for predicting the magnitude of antimalarial effects.

EXPERIMENTAL (48)

Fused 2-Aminothiophene-3-carbonitriles (XI) (1-9, Table I). Procedure I

To a stirred suspension of 8.7 g. (0.05 mole) of 4-phenylcyclohexanone (49), 3.3 g. (0.05 mole) of malononitrile, and 1.6 g. (0.05 mole) of sulfur in 30 ml. of ethanol was added gradually 5 ml. of morpholine. Heat was evolved, and the temperature rose to 43° . Partial solution occurred, but before complete solution was attained a copious precipitate formed. The reaction mixture was stirred at room temperature for 3 hours and was filtered. The product was collected and recrystallized from 180 ml. of ethanol to give 8.0 g. (63%) of 2-amino-4,5,6,7-tetrahydro-6-phenylbenzo[b]thiophene-3-carbonitrile (7) as shiny plates, m.p. $177.5-180.5^{\circ}$.

Procedure II.

A suspension of 54.9 g. (0.28 mole) of 3,4-dihydro- $\Delta^{1}(2H),\alpha$, naphthalenemalononitrile (50) and 9.1 g. of sulfur in 500 ml. of ethanol containing 25 ml. of morpholine was stirred and heated under reflux for 12 hours. The reaction mixture was allowed

77

to stand at room temperature for 3 days. The solid that separated was collected (37.4 g.) and recrystallized from 500 ml. of benzene to give 30.4 g. (48%) of 2-amino-4,5-dihydronaphtho[2,1-b]-thiophene-1-carbonitrile (5) as beige crystals, m.p. 165-168°.

Fused 2,4-Diaminothieno[2,3-d] pyrimidines (10-13, 15-19, Table II). Procedure III.

A mixture of 5.8 g. (0.023 mole) of 2-amino-4,5,6,7-tetrahydro-6-phenylbenzo[b]thiophene-3-carbonitrile (7) and 2.9 g. (0.025 mole) of chloroformamidine hydrochloride (38) in 12 ml. of diglyme was stirred and heated in an oil bath at 150-155°. Hydrogen chloride was evolved and a clear solution formed. After 0.75 hour, a new solid had precipitated and the mixture was cooled. The yellow solid was collected, washed successively with diglyme and ether, and air dried. The solid was suspended in dilute sodium hydroxide, stirred at room temperature for 0.5 hour, collected, washed with water, and dried. Crystallization from dimethylformamide containing dilute sodium hydroxide solution gave yellow crystals of 2,4-diamino-5,6,7,8-tetrahydro-7-phenyl-[1]benzothieno[2,3-d]pyrimidine (17). After drying in vacuo at 50° for 18 hours, the product weighed 4.5 g. (67%), m.p. 267-269°.

Procedure IV.

A suspension of 5.9 g. (0.026 mole) of 2-amino-4,5-dihydronaphtho[2,1-b]thiophene-1-carbonitrile (5) in 15 ml. of diglyme was heated in an oil bath to 130° to effect solution. Chloroformamidine hydrochloride (38) (3.3 g., 0.029 mole) was added and the bath temperature was increased to 155°. After several minutes a clear orange solution formed, hydrogen chloride was evolved, and a yellow solid formed. After 10 minutes, the reaction was cooled and the solid was collected, washed successively with diglyme and ethyl acetate, and dried in the air (5.4 g.). The solid was dissolved in 200 ml. of hot 95% ethanol containing 2 ml. of concentrated ammonium hydroxide, and the mixture was diluted with 350 ml. of water and chilled. The solid that formed was collected, washed with water, and dried in vacuo at 50° to give 4.0 g. (58%) of (1-cyano-4,5-dihydronaphtho[2,1-b]thien-2yl)guanidine (14) as off-white crystals, m.p. 304-307° with prior sintering.

Anal. Calcd. for $C_{14}H_{12}N_4S$: C, 62.66; H, 4.51; N, 20.88. Found: C, 62.54; H, 4.57; N, 21.14.

The ir spectrum showed a strong nitrile band at 2210 cm⁻¹. A solution of 3.5 g. (0.013 mole) of (1-cyano-4,5-dihydronaphtho[2,1-b]thien-2-yl)guanidine (14) in 7.5 ml. of dimethylformamide was heated under reflux for 3 hours. The mixture was cooled, filtered, and the solid was washed with ethanol and air dried. The desired 9,11-diamino-5,6-dihydronaphtho[1',2':4,5]-thiano[2,3-d]pyrimidine (15) (3.0 g., 86%) (50% overall) was obtained as beige needles, m.p. 306-308°.

$\Delta^{1,\alpha}$ Indanmalononitrile.

A solution of 66.0 g. (0.5 mole) of 1-indanone and 33.0 g. (0.5 mole) of malononitrile in 200 ml. of benzene containing 4.0 g. of ammonium acetate and 12 ml. of acetic acid was heated under reflux under a Dean Stark water separator for 9 hours. A total of 13.5 ml. of water was collected. The mixture was cooled and filtered. The solid (70.7 g.) was recrystallized from benzene to give 50.3 g. (56%) of product. An analytical sample was prepared by recrystallization from benzene, m.p. 146-150°.

Anal. Calcd. for $C_{12}\,H_8\,N_2$: C, 79.98; H, 4.47; N, 15.55. Found: C, 79.89; H, 4.76; N, 15.49.

2-Phenyl- Δ^{1} , α -cyclohexanemalononitrile.

A solution of 41.0 g. (0.235 mole) of 2-phenylcyclohexanone and 15.5 g. (0.235 mole) of malononitrile in 200 ml. of benzene containing 1.9 g. of ammonium acetate and 5.7 ml. of acetic acid was heated under reflux utilizing a Dean Stark water separator. After 4 hours, 7.0 ml. of water had collected. The solvent was removed in vacuo and the residual syrup was suspended in water and made basic with aqueous sodium carbonate. The oil was induced to solidify by scratching, and the solid was collected, washed with water and dried in air to give 49.0 g. of crude product. Recrystallization from 200 ml. of 2-propanol yielded 40.2 g. (77%) of product, m.p. 63-67.5°. A small sample was recrystallized again for analysis, m.p. 65-69°.

Anal. Calcd. for $C_{15}H_{14}N_2$: C, 81.05; H, 6.35; N, 12.60. Found: C, 80.92; H, 6.48; N, 12.68.

Acknowledgments.

The authors are indebted to Dr. Leo Rane of the University of Miami and Dr. Karl H. Rieckmann of the University of Chicago for the antimalarial testing. We also thank Dr. M. W. Fisher and Dr. C. L. Heifetz of Parke, Davis and Company for the antibacterial screening, and Dr. C. C. Smith of the University of Cincinnati for the antimetabolite studies. Chemical assistance was supplied by Miss B. Christy and Mr. R. Cooley, microanalyses by Mr. C. E. Childs and associates, and spectral data by Dr. J. M. Vandenbelt and co-workers, all of Parke, Davis and Company.

REFERENCES

- (1) For paper 5 on folate antagonists, see E. F. Elslager, J. Clarke, J. Johnson, L. M. Werbel and J. Davoll, *J. Heterocyclic Chem.*, 9,767 (1972).
- (2) This is communication 28 of a series on antimalarial drugs. For paper 27, see (1).
- (3) This investigation was supported by U.S. Army Medical Research and Development Command Contract DA-49-193-MD-2754. This is contribution No. 1055 to the Army Research Program on Malaria.
- (4) E. F. Elslager, A. Curry and L. M. Werbel, J. Heterocyclic Chem., 9, submitted for publication.
- (5) E. F. Elslager, J. Clarke, P. Jacob, L. M. Werbel and J. D. Willis, *ibid.*, 9, submitted for publication.
- (6) J. Davoll, A. M. Johnson, H. J. Davies, O. D. Bird, J. Clarke and E. F. Elslager, J. Med. Chem., 15, 000 (1972).
- (7) P. E. Thompson, A. Bayles and B. Olszewski, Exp. Parasitol., 25, 32 (1969).
- (8) J. Davoll, J. Clarke and E. F. Elslager, J. Med. Chem., 15, 000 (1972).
- (9) J. Davoll, A. M. Johnson, J. Dickinson, O. D. Bird, J. Clarke and E. F. Elslager, *ibid.*, 16, 000 (1973).
- (10) P. E. Thompson, A. Bayles and B. Olszewski, Am. J. Trop. Med. Hyg., 19, 12 (1970).
- (11) E. F. Elslager, O. D. Bird, J. Clarke, S. C. Perricone, D. F. Worth and J. Davoll, J. Med. Chem., 15, 000 (1972).
- (12) E. F. Elslager, J. Davoll, L. M. Werbel and D. F. Worth, Abstracts of Papers, Third International Congress of Heterocyclic Chemistry, Sendai, Japan, August 23-27, 1972, pp. 366-369.
- (13) E. F. Elslager, J. Clarke, L. M. Werbel, D. F. Worth and J. Davoll, J. Med. Chem., 15, 000 (1972).
- (14) B. Roth, ibid., 12, 227 (1969).
- (15) J. J. Burchall and G. H. Hitchings, *Mol. Pharmacol.*, 1, 126 (1965).
- (16) P. Schmidt, K. Eichenberger and E. Schweizer, German Patent 1,908,497, September 18, 1969; *Chem. Abstr.*, 72, 31837u (1970).

- (17) K. Eichenberger, P. Schmidt and E. Schweizer, U.S. Patent 3,627,766, December 14, 1971.
- (18) E. A. Falco, L. G. Goodwin, G. H. Hitchings, I. M. Rollo and P. B. Russell, *Brit. J. Pharmacol.*, 6, 185 (1951).
- (19) G. H. Hitchings and J. J. Burchall, "Advances in Enzymology," Vol. XXVII, Interscience Inc., New York, N.Y., 1965, pp. 417-468.
- (20) A. Rosowsky and E. J. Modest, Ann. N.Y. Acad. Sci., 186, 262 (1971).
- (21) A. Rosowsky, M. Chaykovsky, K. K. N. Chen, M. Lin and E. J. Modest, Abstracts of Papers, 163rd ACS National Meeting, Boston, Mass., April 11, 1972, MEDI 33.
- (22) A. Rosowsky, M. Chaykovsky, K. K. N. Chen, M. Lin and E. J. Modest, *J. Med. Chem.*, 15, 000 (1972).
- (23) M. Chaykovsky, M. Lin, A. Rosowsky and E. J. Modest, ibid., 15, 000 (1972).
- (24) A. Rosowsky, K. K. N. Chen and M. Lin, *ibid.*, 15, 000 (1972).
- (25) B. R. Baker, J. P. Joseph, R. E. Schaub, F. J. McEvoy and J. H. Williams, J. Org. Chem., 19, 138 (1953).
- (26) A. A. Santilli, D. H. Kim and S. V. Wanser, J. Heterocyclic Chem., 8, 445 (1971), and references cited therein.
- (27) A. M. Chacko, Ph.D. Dissertation, University of North Carolina at Chapel Hill, 1965; Univ. Microfilms, 65-14320, Ann Arbor, Michigan; *Dissertation Abstr.*, 26, 3627 (1966).
- (28) E. C. Taylor and A. McKillop, "The Chemistry of Cyclic Enaminonitriles and o-Aminonitriles," in "Advances in Organic Chemistry," Vol. 7, Interscience Publishers, New York, N.Y., 1970.
- (29) E. C. Taylor and J. G. Berger, Angew. Chem., 78, 144 (1966); Int. Ed. Engl., 5, 131 (1966).
- (30) E. C. Taylor and J. G. Berger, *J. Org. Chem.*, 32, 3276 (1967).
- (31) F. Sauter, Monatsh. Chem., 99, 1507, 2100 (1968); 101, 535 (1970).
- (32) L. Capuano, M. Welter and R. Zander, Chem. Ber., 102, 3698 (1969).
- (33) W. O. Foye, J. Mickles and G. M. Boyce, J. Pharm. Sci., 59, 1348 (1970).

- (34) F. Asinger, W. Schafer, K. Halcour, A. Saus and H. Triem, Angew. Chem. Int. Ed. Engl., 3, 19 (1964).
- (35) K. Gewald, E. Schinke and H. Bottcher, Chem. Ber., 99 94 (1966).
- (36) Yoshitomi Pharm. Ind. Ltd., German Patent 1,812,404, August, 14, 1969.
- (37) M. Nakanishi, H. Imamura, Y. Maruyama and H. Hoshino, J. Pharm. Soc. Japan, **90**, 272 (1970), ff. 277, 284, 291, 296, 302 (1970).
 - (38) A. Hantzsch and A. Vagt, Ann., 314, 366 (1900).
- (39) A. M. Patterson, L. T. Capell and D. F. Walker, "The Ring Index," 2nd ed., American Chemical Society, Washington, D.C., 1960 and supplements I (1963), II (1964), and III (1965).
- (40) Parenteral antimalarial screening against *P. berghei* in mice and *P. gallinaceum* in chicks was carried out by Dr. Leo Rane of the University of Miami, and test results were supplied through the courtesy of Dr. T. R. Sweeney and Dr. E. A. Steck of the Walter Reed Army Institute of Research.
- (41) For a description of the test method, see T. S. Osdene, P. B. Russell and L. Rane, J. Med. Chem., 10, 431 (1967).
- (42) In vitro studies utilizing P. falciparum were carried out by Dr. Karl Rieckmann of the University of Chicago. Test results were supplied through the courtesy of Dr. T. R. Sweeney and Dr. E. A. Steck of the Walter Reed Army Institute of Research.
- (43) For a description of test procedures, see K. H. Rieckmann, J. V. McNamara, H. Frischer, T. A. Stockert, P. E. Carson and R. D. Powell, Am. J. Trop. Med. Hyg., 17, 661 (1968).
- (44) K. H. Rieckmann, J. Amer. Med. Ass., 217, 573 (1971).
- (45) W. Szybalski, Microbial Genetics Bull., 5, 16 (1951).
- (46) A. H. Webb and L. Washington, Bacteriol. Proc., 52 (1966).
- (47) Antimetabolite studies were conducted by Dr. Carl C. Smith of the University of Cincinnati, and test results were supplied through the courtesy of Dr. T. R. Sweeney and Dr. E. A. Steck of the Walter Reed Army Institute of Research.
- (48) Melting points (corrected) were taken on a Thomas-Hoover capillary melting point apparatus.
 - (49) H. E. Ungnade, J. Org. Chem., 13, 361 (1948).
- (50) D. T. Mowry, J. Am. Chem. Soc., 67, 1050 (1945).